



Oklahoma Enhanced Aquifer Recharge (EAR) Project

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GWPC ASR-MAR Webinar

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External Partners

- Chickasaw Nation
- City of Ada, OK
- Oklahoma State University
- East Central University
- U.S. Geological Survey
- Oklahoma Water Resources Board
- Oklahoma Department of Environmental Quality
- EPA Region 6





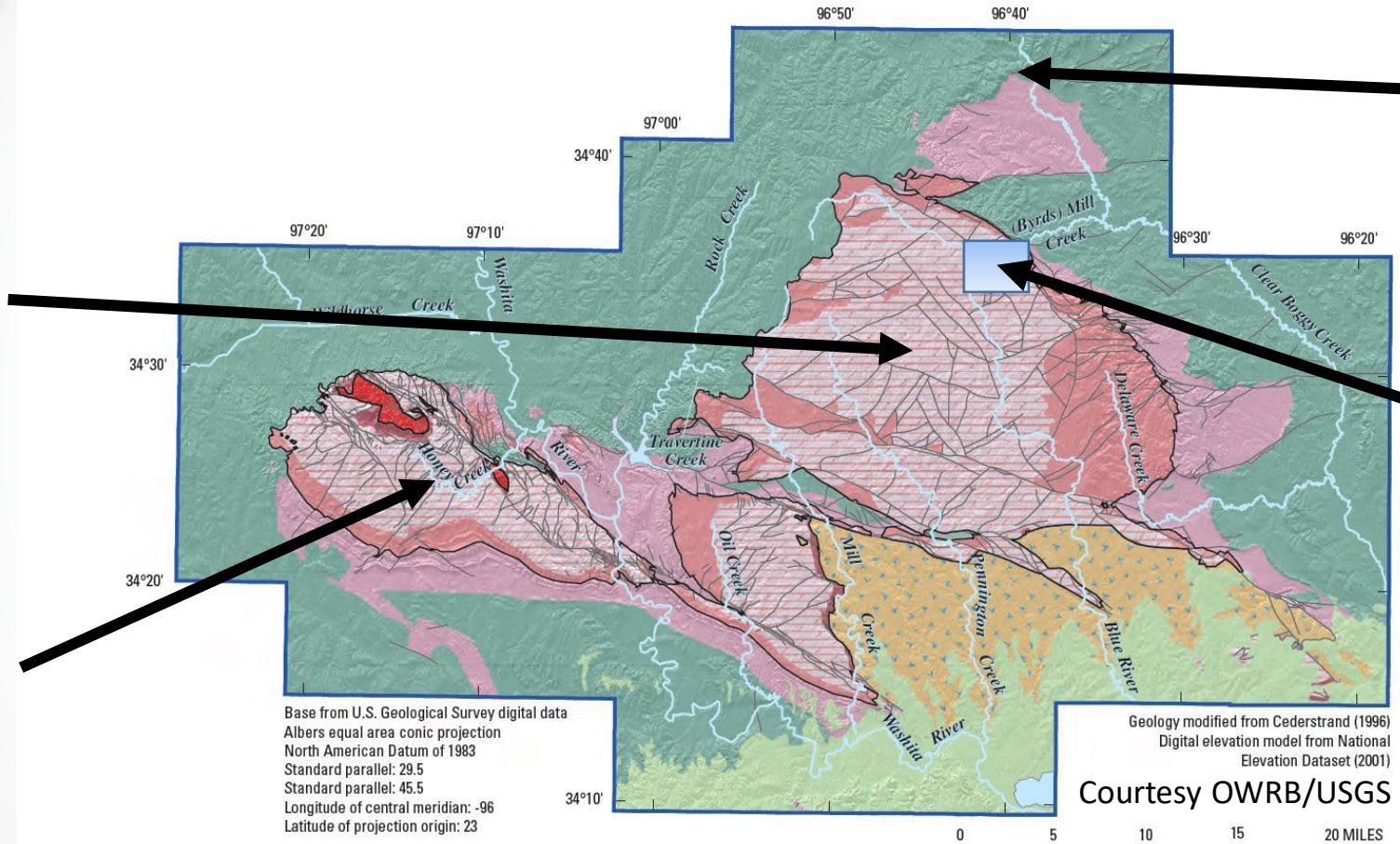
Arbuckle-Simpson Aquifer Geology

Hunton Anticline

Arbuckle Mountains

Ada
(GCRD)

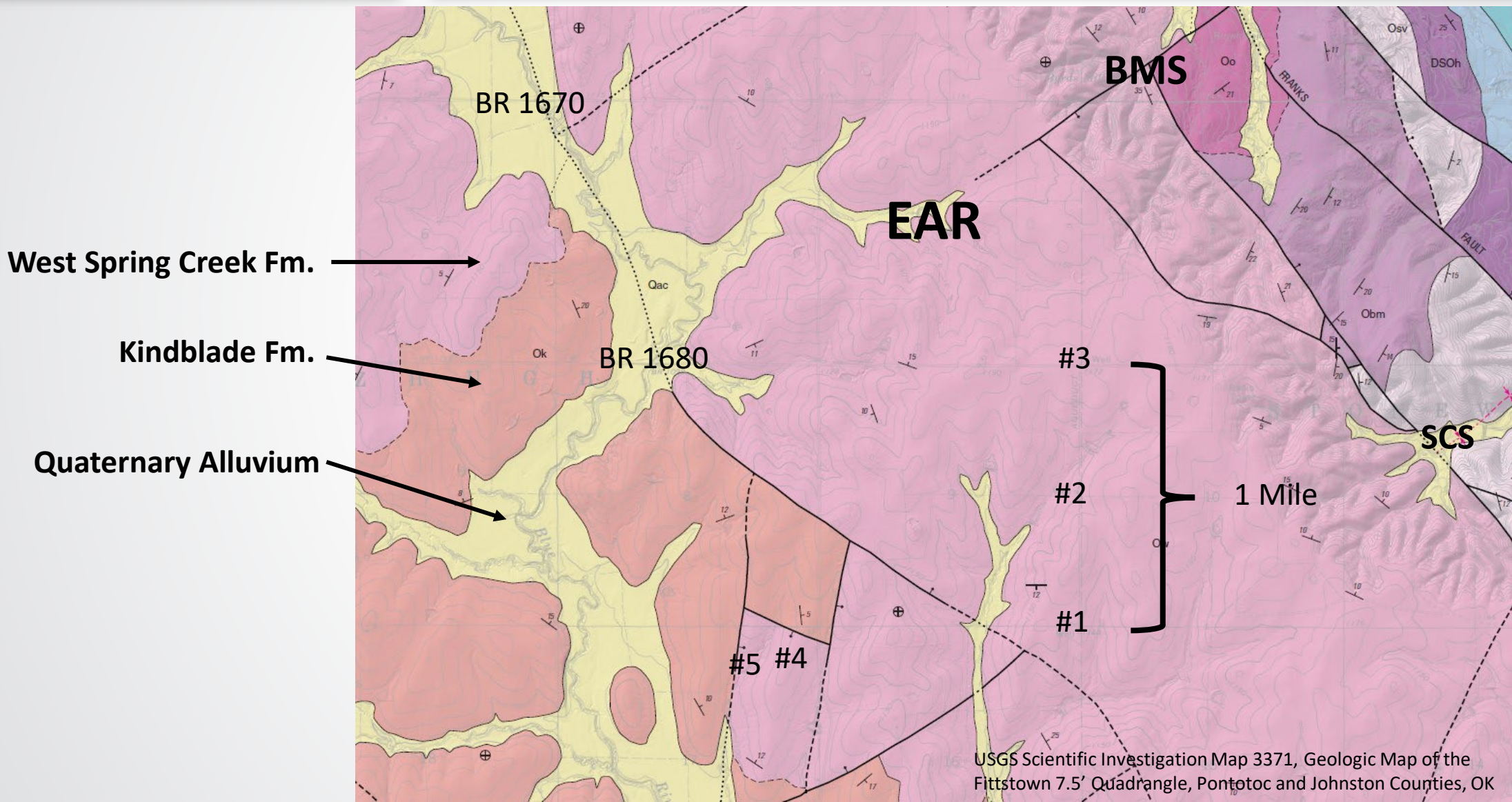
EAR
Study
Area



EXPLANATION

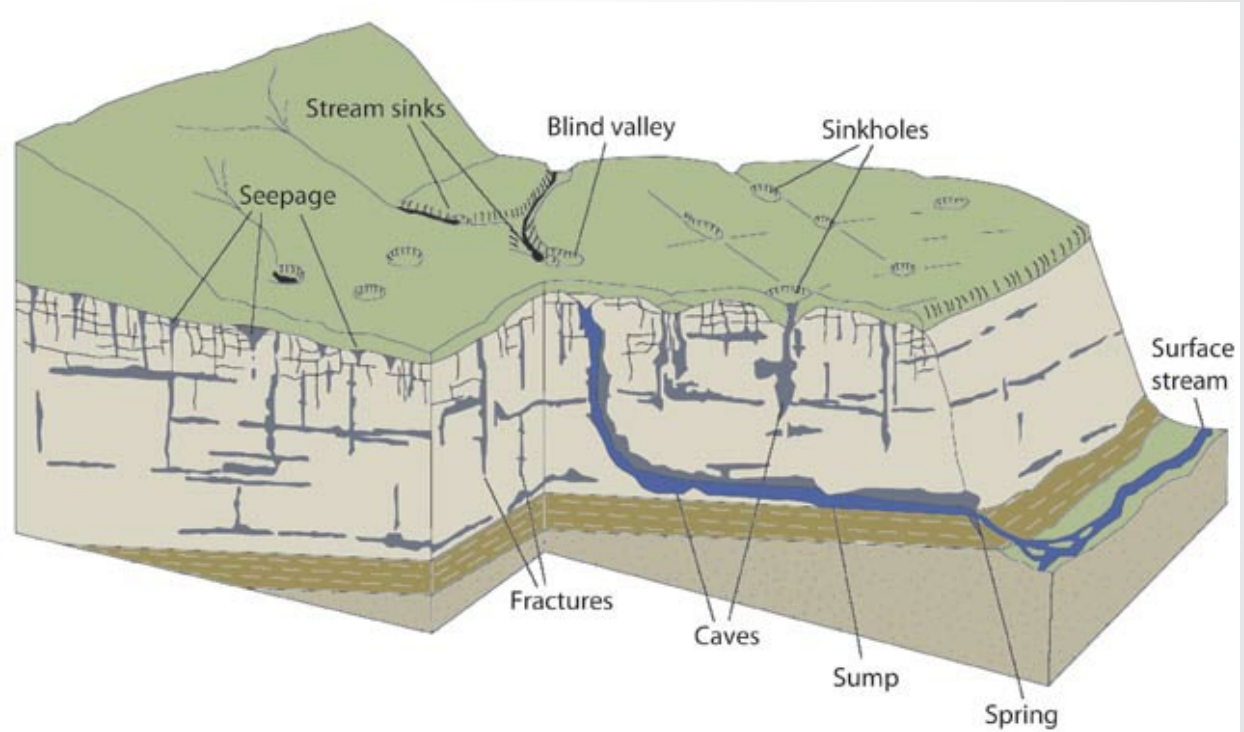
Rock-stratigraphic unit					
	Cretaceous formation		Ordovician-Cambrian Arbuckle Group		Water
	Permian-Pennsylvanian formations		Cambrian Timbered Hills Group		Fault
	Mississippian-Ordovician formations		Cambrian Colbert Rhyolite		Boundary of study area
	Ordovician Simpson Group		Proterozoic igneous and metamorphic rocks		Arbuckle-Simpson aquifer boundary

EAR Area Geology

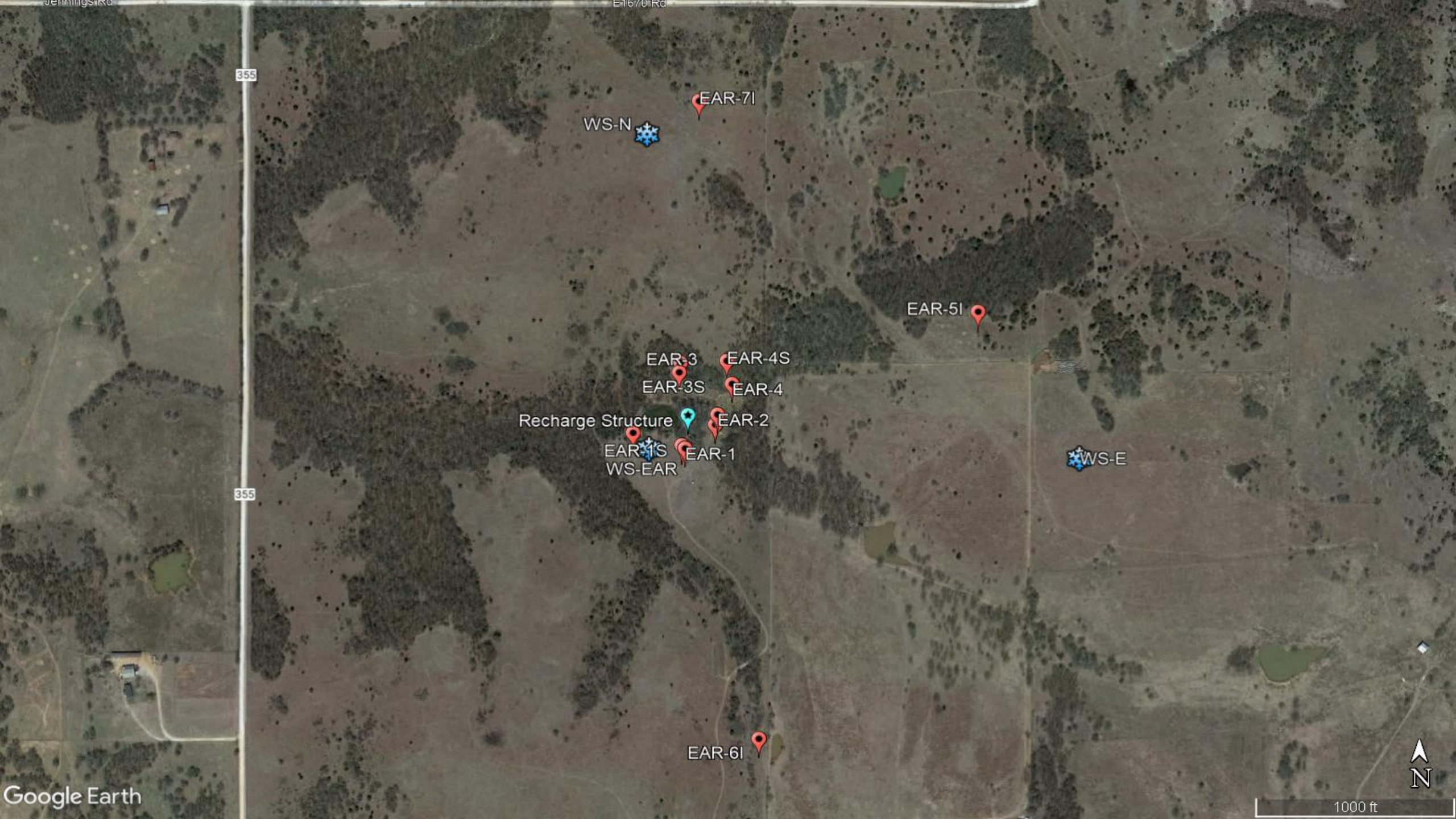


Karst Hydrogeology

- **Primarily composed of carbonates (i.e., limestone and dolomite)**
- **Preferential flow paths develop through dissolution and expansion of faults, fractures, bedding planes, etc.**
- **Groundwater travel times vary orders of magnitude (days to years)**



Source: Wisconsin Geological and Natural History Survey, 2021



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WS-N

EAR-71

EAR-51

EAR-3

EAR-4S

EAR-3S

EAR-4

Recharge Structure

EAR-2

EAR-1S

EAR-1

WS-EAR

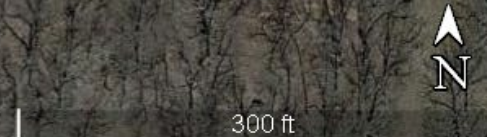
WS-E

355

EAR-61



1000 ft





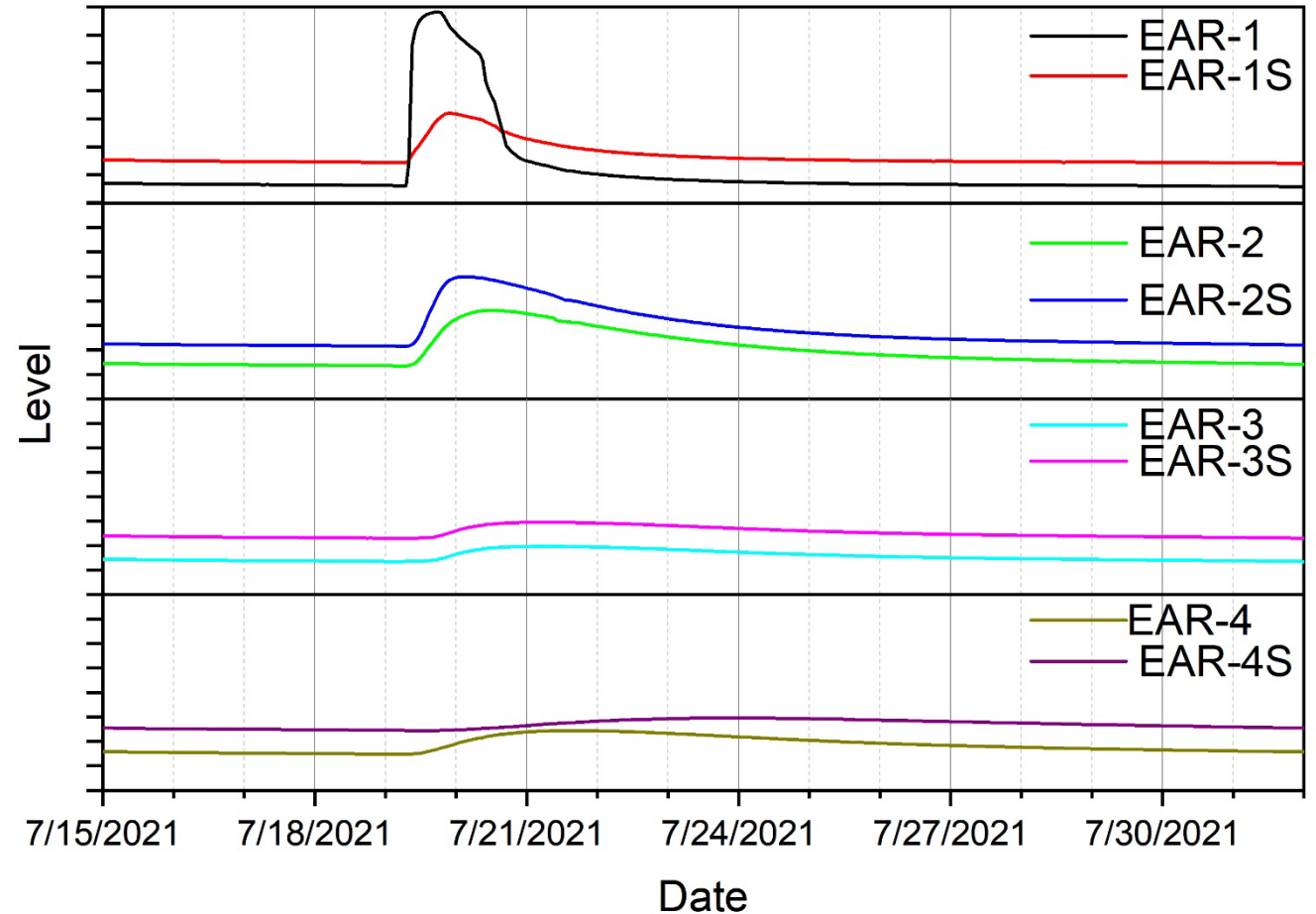
- Sinkhole estimated recharge rate ~1400 gpm based on volumetric calculations and water level declines (e.g., pressure transducer data).
- Local water levels rapidly increase after flooding of sinkhole



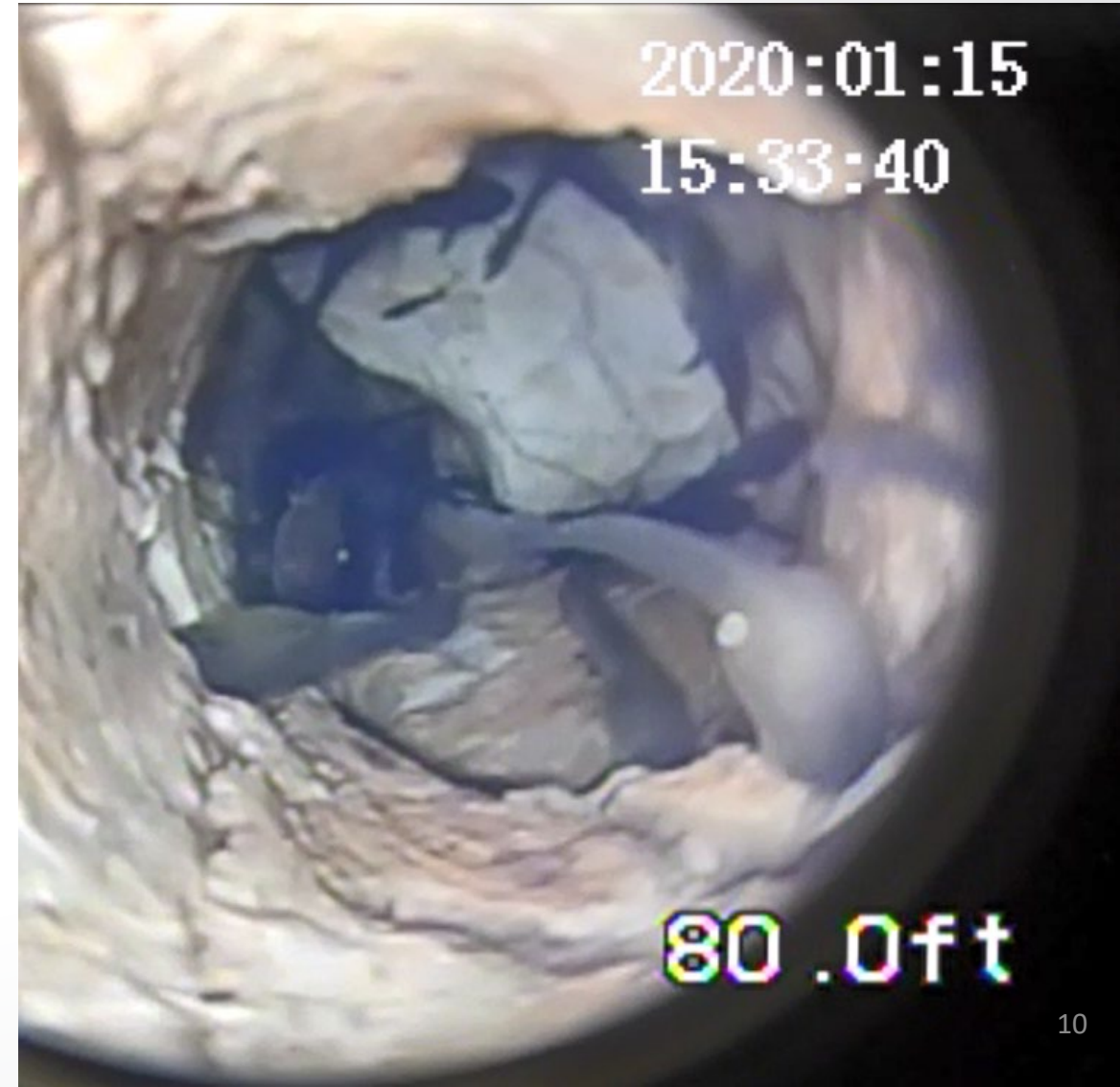


Time To Peak Water levels (July 2021)

Well	Elapsed Time (hr)
EAR-1	11
EAR-1S	14
EAR-2	35
EAR-2S	21
EAR-3	42
EAR-3S	42
EAR-4	59
EAR-4S	123

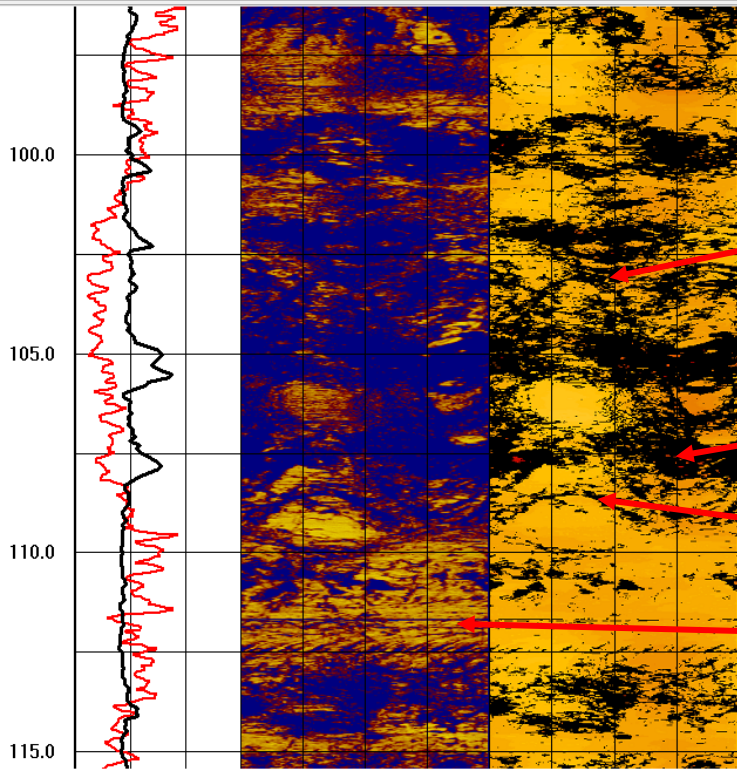
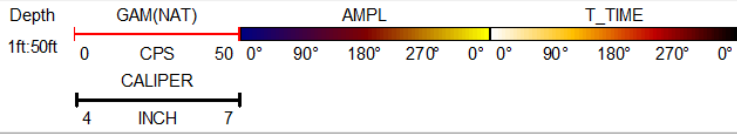


- Rapid water level response to overland flow events.
- Direct connection between sinkhole (i.e., fish & tadpoles).





EAR Site Geophysical Logs



Fracture Trace

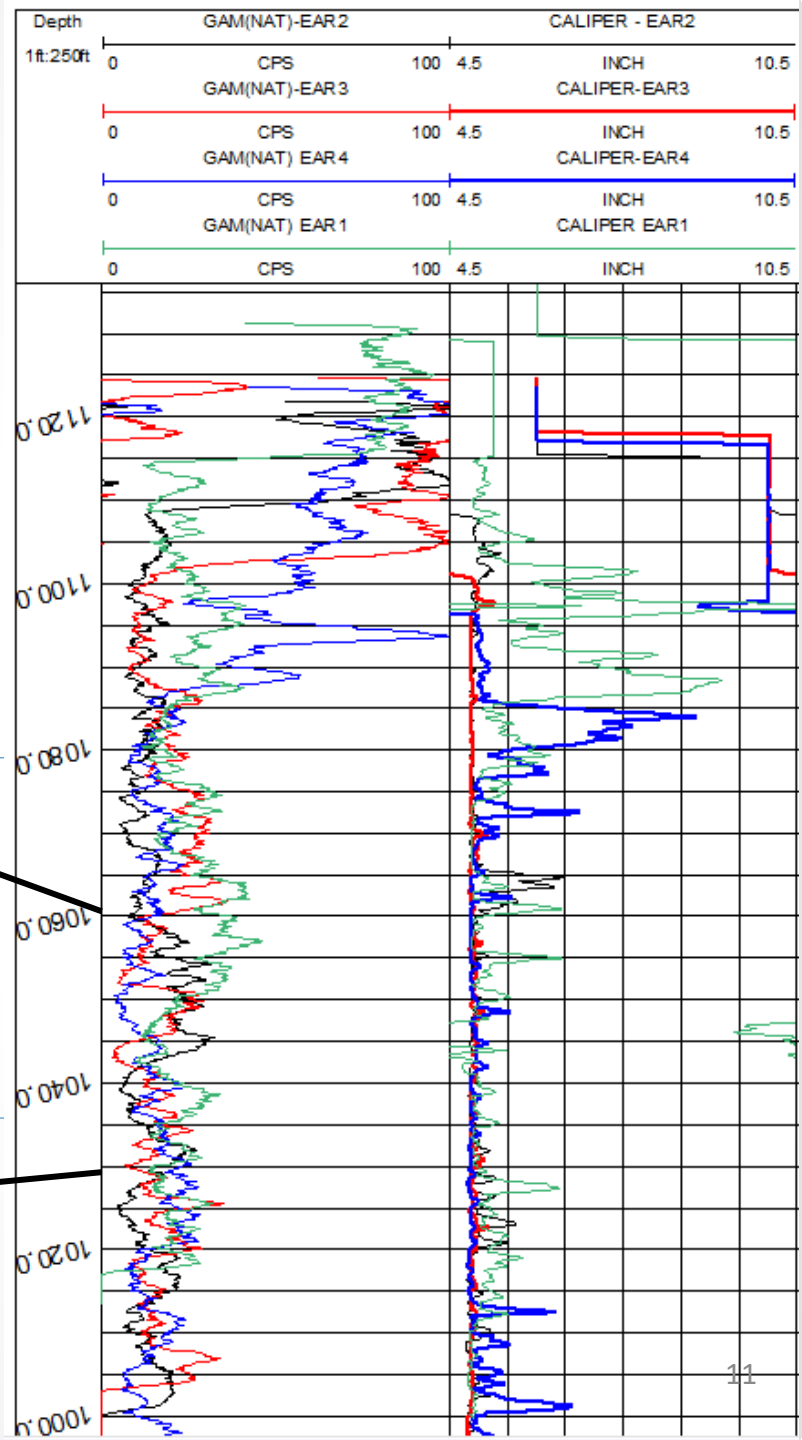
Dissolution Feature

Fracture Trace

Drill Bit Grooves
(Competent Rock)

High Water Level

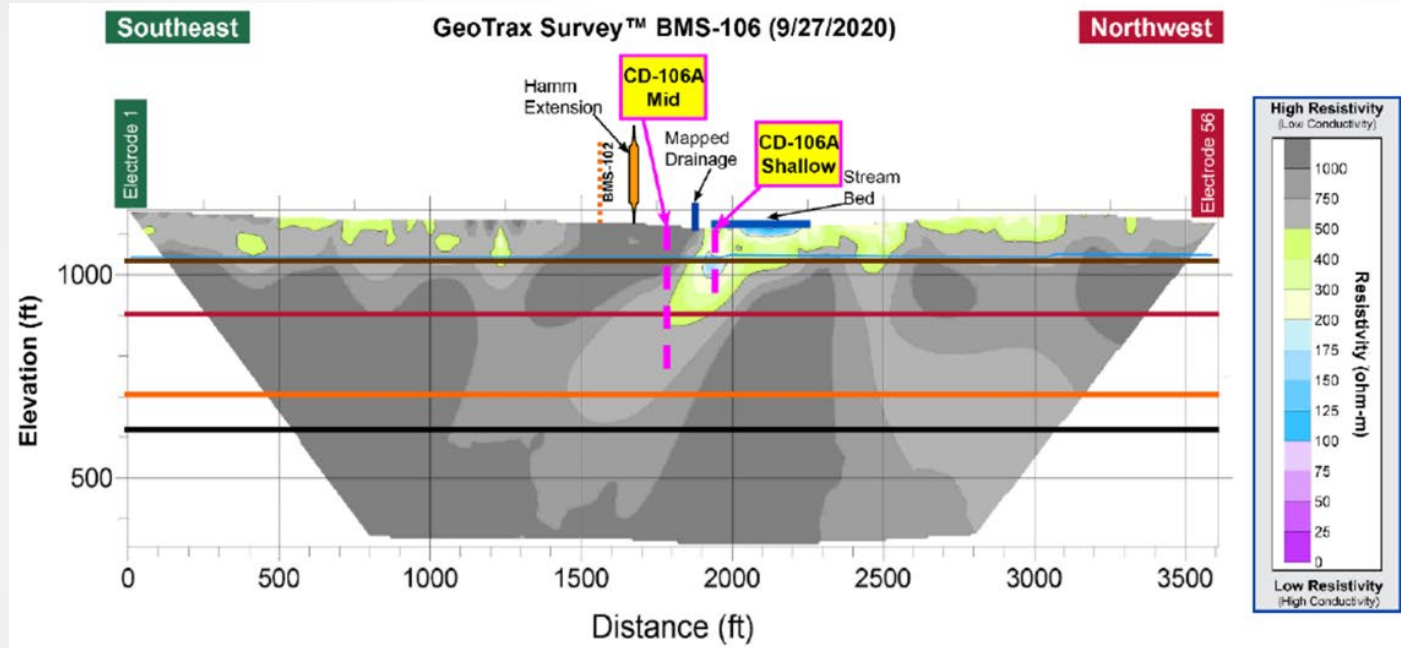
Low Water Level



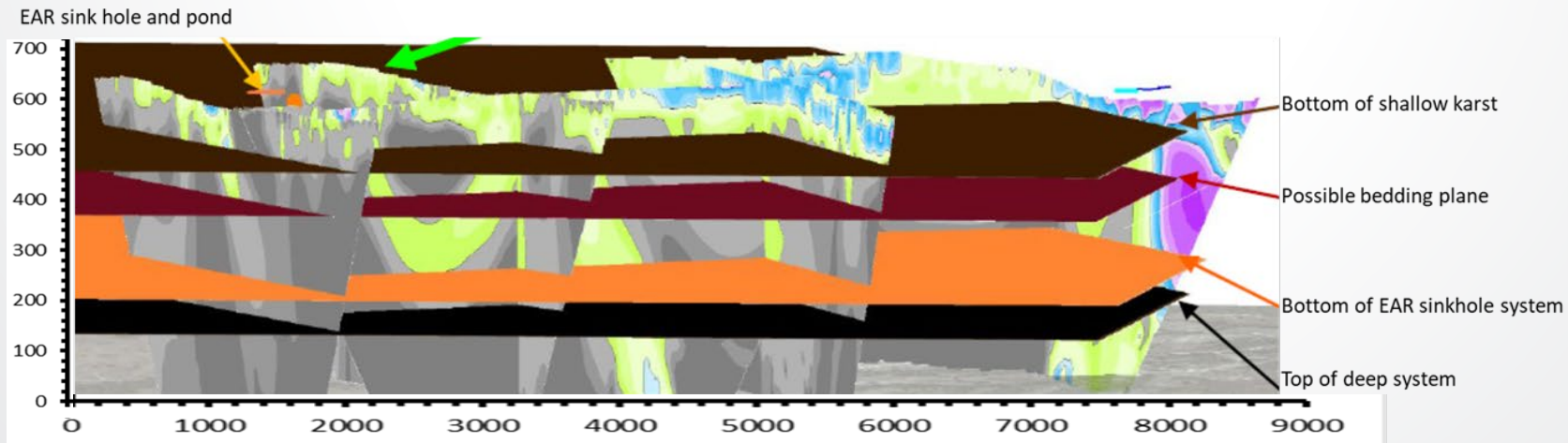
EAR-2 Acoustic Televiwer (ATV), Caliper
& Natural Gamma Logs



ERI Investigations - Deep Hydrology ("Plumbing")



Figures from Aestus, Inc., 2020



Sample Types

- Groundwater
 - Wells
 - Springs
- Surface water
 - Blue River
 - Pond
- Precipitation
- Soil porewater (vadose zone)
- Climate data
- Soil/ sediment/ rock cores
- Soil moisture





Parameters Analyzed

- Field parameters
 - T, SPC, TDS, DO, pH, ORP, turbidity, alkalinity
- Major anions and cations
 - Ca^{2+} , Mg^{2+} , K^+ , Na^+ , Cl^- , SO_4^{2-} , HCO_3^- , CO_3^{2-}
- Nutrients
 - NO_3^- , NH_3 , PO_4^{3-} , TN, TP, DOC, TOC
- Trace elements
 - F^- , I^- , Trace metals
- Isotopes
 - Water isotopes ($\delta^{18}\text{O}$ & $\delta^2\text{H}$)
 - Strontium Isotopes ($^{87}\text{Sr}/^{86}\text{Sr}$, $^{88}\text{Sr}/^{86}\text{Sr}$)
- Volatile organic compounds
- Dissolved gases
 - CO_2 , CH_4 , N_2O
- Microbial
 - Coliforms, *E. coli*, Enterococci
 - DNA



MAR-EAR Implementation Needs

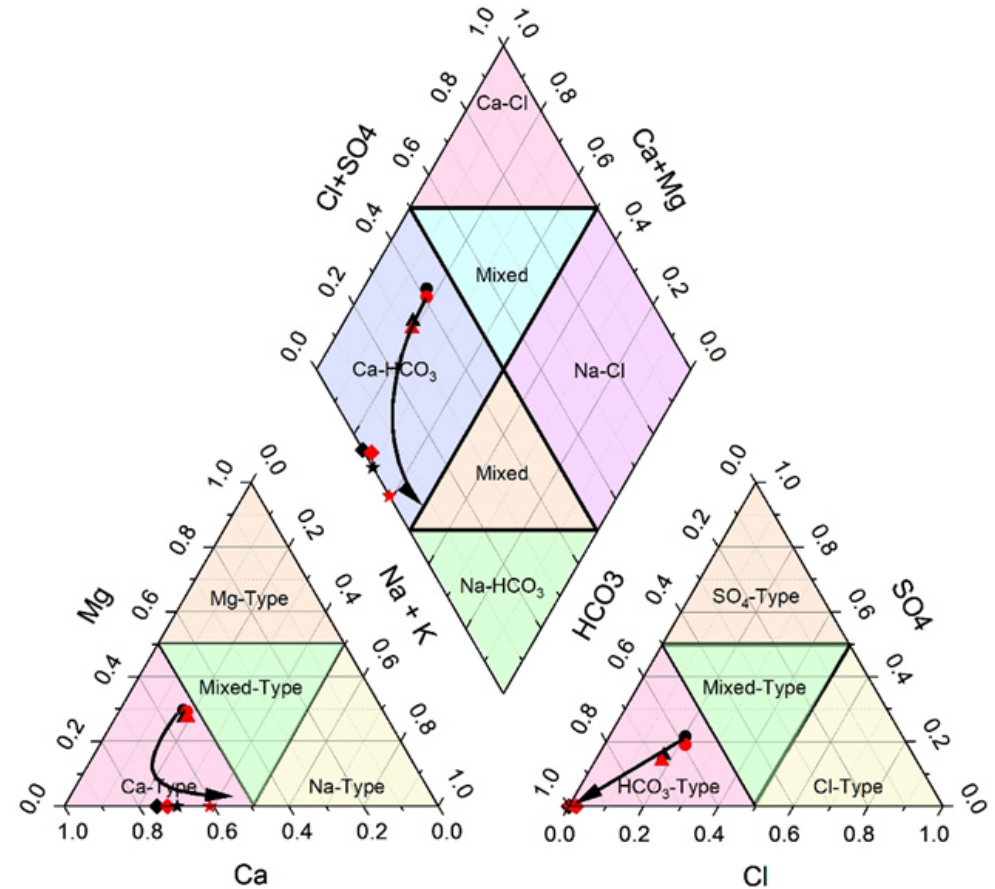
- Site Characterization
 - More than just hydrological considerations
 - Assess ambient groundwater
 - Assess the “chemical geology”
 - What are the minerals that make up the aquifers and vadose zone (likely heterogenous distribution)
 - What are the potential natural contaminants present in system
 - What are the potential biogeochemical processes
 - What conditions would cause the release of naturally contaminants



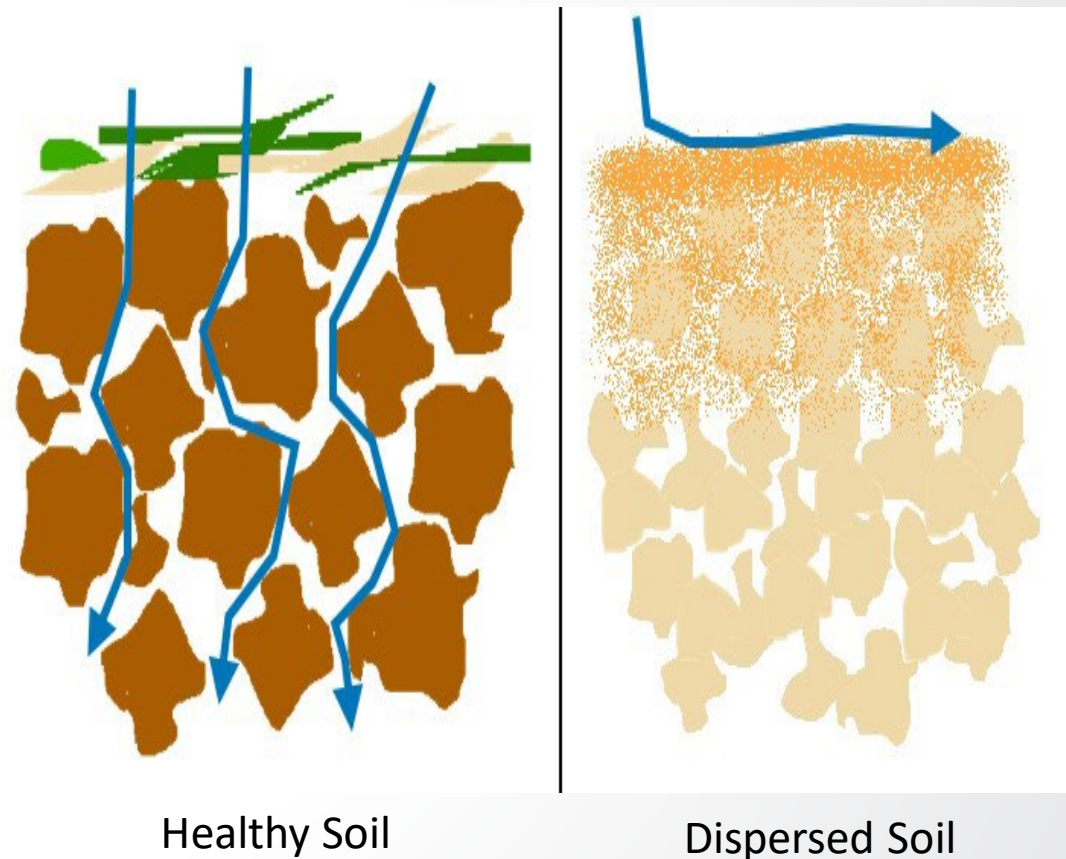
MAR-EAR Implementation Needs Continued

- Characterization of source water
 - How will the source water “interact” with native water
 - Does the source water contain contaminants of concern?
 - Will source water potentially mobilize naturally occurring contaminants present?
 - Will source water chemically react with native water and produce contaminants of concern?
 - Will the mixing of the source water with the ambient groundwater produce potential drinking water treatment impacts?
 - Assess potential source water treatment/ mitigation strategies
- Long term monitoring
 - Assess water quality
 - Assess performance

- Mixing and dilution of infiltrated stormwater with GW caused changes to GW quality
- Concentrations of major anions and cations in stormwater <<< GW
 - Except when de-icing agents applied
- If GW is used for drinking water:
 - Dilution and increasing sodium can cause water type changes
 - Impacts to drinking water treatment are possible



- De-icing can impact water quality and quantity
- Data suggested reverse ion exchange was occurring in vadose zone
 - Loads sodium onto particle surface (e.g., clays)
 - Dispersion of particles
 - Reverse ion exchange can lead to:
 - Clogging
 - Diminished infiltration and water movement
- Enhanced mobility of some metals could be problematic when chloride salts are applied as de-icing agents and stormwater is infiltrated



Question?

